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SYNTHESIS AND BIOLOGICAL EVALUATION OF SOME ACYCLIC NUCLEOSIDE CYCLIC PHOSPHORAMIDATE DERIVATIVES

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Abstract

The acyclic nucleosides 2 were treated with 2-chloro-3-methyl-1-oxa-3-aza-2-phosphacyclopentane (3) in the presence of disopropylethylamine to give the corresponding phosphoramidite derivatives (4). The phosphoramidite intermediates (4) were oxidized with m-chloroperbenzoic acid to the phosphoramidate derivatives (5). Treatment of 5a,b with ZnBr₂ in CH₃NO₂ gave the corresponding acyclic nucleoside cyclic phosphoramidates (6a,b). Attempts to desilylation of 5c by tetrabutylammonium fluoride (TBAF) resulted in opening of the phosphoramidate ring. The newly synthesized compounds were evaluated for antiviral and antitumor cell activity.

INTRODUCTION

Recently, several workers reported the chemical syntheses and biological evaluation of nucleoside cyclic phosphoramidate derivatives. These compounds are hydrolyzed under chemical or enzymic conditions and are a potential source of nucleoside 5'-phosphates in vivo. On the other hand, acyclic nucleoside analogues have gained increasing interest because of their biological activity, particularly as antiviral and anticancer compounds (1).²

We now report the synthesis of acyclic nucleoside phosphoramidate derivatives $(\underline{6})$ which were evaluated for their antiviral and antitumor cell activity.

CHEMISTRY

The tritylation or silvlation of acyclic nucleoside $(\underline{1})$ was carried out as follows: the acyclic nucleoside l was treated with dimethoxytri-

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tyl chloride (1.1 molar equiv.) in dry pyridine for 5 h to give the corresponding tritylated compounds $\underline{2a}$ (56 %) and $\underline{2b}$ (57 %), respectively along with bis-tritylated compounds, after separation by silica gel column chromatography. While treatment of $\underline{1a}$ with tert-butyldimethylsilyl chloride (TBDMS-Cl) (1.1 molar equiv.) in dry DMF gave the silylated product (2c) (47 %).

Next we examined the phosphorylation of 2 by using 2-chloro-3-methyl-l-oxa-3-aza-2-phosphacyclopentane (3). The phosphitylating agent 3 was prepared according to the procedure reported by Martynov. 3 However, we observed that the phosphitylating agent 3 was lower than 90 % pure (by 31 P-NMR). Satisfactory 31 P-NMR data of 3 were obtained only after distillation over cesium fluoride. The phosphitylating agent 3 (1.5 molar equiv.) thus obtained was treated with 2a in the presence of disopropylethylamine (1.5 molar equiv.) in dry THF at -50°C. After 30 min, the precipitated diisopropylethylammonium hydrochloride was removed by filtration and the filtrate was treated with m-chloroperbenzoic acid (MCPBA) at 20°C for 15 min. 4 After the usual work up, the corresponding acyclic nucleoside cyclic phosphoramidate (5a) was isolated in 79 % yield. As a more direct method for preparing 5a we examined the direct condensation reaction of 2-chloro-3-methy1-1-oxa-3-aza-2-phosphacyclopentane 2-oxide (7) with an acyclic nucleoside. However, compound 7 failed to react with $\underline{3}$ under several conditions. Compound $\underline{5a}$ was characterized by UV and NMR spectroscopy and elemental analysis. The NCH $_{\rm q}$ signal of the $^{\rm l}$ H-NMR spectrum consists of a pair of doublets with a phosphorus coupling constant of 10 Hz. The further splitting of these signals is ascribed to the presence of diastereomers in the sample resulting from a chiral phosphorus centre. Two signals (22.63 and 22.29 ppm) observed in the 31 P-NMR spectrum could be assigned to the diastereomers, present in the ratio 2:1.

In a similar manner, the compounds 5b,c were obtained in 76 % and 79 % yield, respectively. The $^1{\rm H}$ and $^{31}{\rm P-NMR}$ spectra of these products were consistent with the assigned structures and showed the presence of diastereomers.

Finally, we examined the possibility of removing the hydroxyl protecting groups from $\underline{5}$. Compound $\underline{5c}$ was treated with TBAF at low temperature. After 2 h, thin-layer chromatography (TLC) analysis (solvent B)

showed complete conversion of $\underline{5c}$ (Rf 0.38) into base line material. From this result, the desilylation by TBAF treatment was accompanied by opening of the phosphorus heterocycle. To obtain compounds $\underline{6a,b}$ from $\underline{5a,b}$, the reaction with ZnBr_2 was carried out in $\mathrm{CH_3NO_2}$. This gives selective removal of the hydroxyl protecting group from $\underline{5}$. The compounds had the expected characteristic UV and NMR spectra.

BIOLOGICAL ACTIVITY

Compounds <u>la</u>, <u>lb</u>, <u>6a</u>, <u>6b</u>, <u>8a</u> and <u>8b</u> were evaluated for antiviral activity, cytotoxicity and antitumor cell activity according to well established procedures. No cytotoxicity or antiviral activity was observed for any of these compounds under conditions where the reference compounds tubercidin, (\underline{S})-DHPA [(\underline{S})-9-(2,3-dihydroxypropyl)adenine], ribavirin and carbocyclic 3-deazaadenosine showed the usual cytotoxicity or antiviral activity (Table 1). Also, compounds <u>la</u> and <u>6a</u> were totally devoid of antitumor cell activity (Table 2). Compounds <u>lb</u>, <u>6b</u>, <u>8a</u> and <u>8b</u> did show some antitumor cell activity, but in terms of potency, they did not even come close to the antitumor cell potency exhibited by the standard fluorinated antitumor agents, 5-fluorouracil, 5-fluorouridine and 5-fluoro-2'-deoxyuridine (Table 2).

EXPERIMENTAL SECTION

Ultraviolet spectra were recorded on a Shimazu UV 200 spectrometer.

H-NMR spectra were recorded in a JOEL JNMPS 100 spectrometer with TMS internal standard. Thin layer chromatography (TLC) was performed on precoated TLC plates of silica gel 60 F-254 (Merck, Art. No. 5715) and the R_f values of the acyclic nucleoside and acyclic nucleotide derivatives were determined after development with solvent A (CH₂Cl₂-MeOH, 95:5, v/v) and solvent B (ethyl acetate-EtOH-pyridine, 8:1:1, v/v). Column chromatography was performed with Silica gel C-200 purchased from Wako Co. Ltd. Pyridine was distilled twice from p-toluenesulfonyl chloride and from CaH₂ and then stored over molecular sieves 4Å. Compounds la,b and 8a,b were prepared according to published procedures.

<u>Tritylation of 1.</u> Compound <u>1a</u> (430 mg, 2 mmol) was treated with dimethoxytrityl chloride (758 mg, 2.2 mmol) in dry pyridine (40 ml) at room temperature for 5 h. The reaction mixture was quenched with ice-water (20 ml) and repeatedly extracted with CH_2Cl_2 (2 x 40 ml). Combined organic

COMPOUNDS

TABLE 1 Cytotoxicity and Antiviral Activity in Primary Rabbit Kidney Cells

Compound	Minimum	Minimal i	lnh1bitory	concentrati	lon ^b (µg/m1)
	cytotoxic concentration ^a (µg/ml)	Herpes simplex virus-1 (KOS)	Herpes simplex virus-2 (G)	Vaccinia virus	Vesicular stomatitis virus
<u>la, lb, 6a, 6b,</u> <u>8a, 8b</u>	> 400	> 400	> 400	> 400	> 400
Tubercidin	> 0.4	> 0.1	> 0.1		0.07
(S)-DHPA	> 400	> 400	> 400	70	40
Ribavirin	> 400	> 400	> 400	10	300
Carbocyclic 3-deazaadenosi	> 400 ne	> 400	> 400	2	1

In HeLa Cells

Compound	Minimal	Minimal inhibit	tory concentration) (µg/ml)
	cytotoxic concentration ^a (µg/ml)	Vesicular stomatitis virus	Coxsackie virus B4	Polio virus-1
<u>la, lb, 6a, 6b,</u> <u>8a, 8b</u>	> 400	> 400	> 400	> 400
Tubercidin (S)-DHPA Ribavirin Carbocyclic 3-deazaadenosi	> 1 > 400 > 200 > 400	0.1 > 400 20 0.7	0.1 > 400 70 > 400	0.2 > 400 70 > 400

In Vero B Cell

Compound	Minimum	Minimum	inhibit	ory conc	entration ^b	(µg/m1)
	cytotoxic concentration ^e (µg/ml)	Para- influenza 3-virus	Reo- virus-l		Coxsackie virus B4	Semliki forest virus
1b	> 400	> 400	150	> 400	> 400	300
1b 1a, 6a, 6b, 8b 8a	> 400	> 200	> 200	> 400	> 400	> 200
<u>8a</u>	> 400	> 100	> 100	100	> 400	> 400
Tubercidin	> 0.4	> 0.1	> 0.1	0.2	0.2	0.2
(S)-DHPA	> 400	20	150	> 400	150	300
Ribavirin	> 400	70	70	70	200	> 400
Carbocyclic 3-deazaadenosi	> 400 ne	0.7	1	70	1	> 200

 $^{^{\}mathbf{a}}$ Required to cause a microscopically detectable alteration of normal cell morphology.

Required to reduce virus-induced cytopathogenicity by 50 %.

Inhibitory Effects on the Proliferation of Tumor Cells $^{\rm a}$ TABLE 2

Compound				10_{50}^{b} (µg/ml)			
	11210/0	L1210/BdUrd	Raji/O	Raj1/TK	FM3A/0	FM3A/TK	Molt/4F
la	> 1000	> 1000	> 1000	> 1000	> 1000	> 1000	ON
의	13.4 ± 3.6	24.5 ± 2.5	353 ± 25	79.3 ± 8.9	2.51 ± 0.81	4.14 ± 0.31	Œ
<u>6a</u>	> 1000	ND ^C	> 1000	ND	648 + 145	QN	> 1000
q 9	205 ± 39	ND	244 ± 20	ND	189 ± 61	ND	212 ± 28
<u>8a</u>	210 ± 28	260 ± 3	63.5 ± 17	38.8 ± 4.1	68.3 ± 27	93.5 ± 38	ON
81	> 1000	627 + 76	26.6 ± 1.1	84.3 ± 42	945 + 72	243 ± 122	QN
5-F-Ura	0.340 ± 0.023	0.451 ± 0.012	9.25 ± 1.66	3.8 + 1.1	0.164 ± 0.038	0.099 ± 0.013	ND
5-F-dUrd	0.00365 ± 0.00030	4.27 ± 0.27	0.178 ± 0.029	1.57 ± 0.83	0.00188 ± 0.00065	0.810 ± 0.180	0.0848 ± 0.0489
5-F-Urd	0.0350 ± 0.0018	0.0354 ± 0.0049	0.126 ± 0.024	0.094 ± 0.024	0.0088 ± 0.0037	0.0246 ± 0.0036	QN
Ado	558 + 55	505 ± 20	350 + 50	792 ± 131	790 ± 280	370 ± 5	ND

^aAbbreviations: L1210/0, murine leukemia cells (wild type); L1210/BdUrd, murine leukemia cells (deoxythymidine kinase-deficient); Raji/TK, human B-lymphoblast cells (deoxythymidine kinase-deficient); FM3A/0, murine mammary carcinoma cells (deoxythymidine kinase-deficient); Molt/4F, human T-lymphoblast cells.
^b50 %-inhibitory dose + standard deviation.
CNot determined.

extracts were washed with water (2 x 20 ml), and the $\mathrm{CH_2Cl_2}$ was evaporated in vacuo. The residue was dissolved in $\mathrm{CH_2Cl_2}$ and purified by silical gel column chromatography. The appropriate fractions [eluted with $\mathrm{CH_2Cl_2}$ -MeOH (95:5, v/v)] were evaporated to ca. 3-4 ml and poured into hexane (100 ml). A white precipitate was collected to give $\underline{2a}$ (560 mg, 56 %); mp 83-85°C; R_f 0.13 (solvent A); UV (MeOH) λ_{max} 265, 254 nm, λ_{min} 234 nm, $\lambda_{\mathrm{H-NMR}}$ (DMSO-d₆) δ 11.37 (br s, 1H, NH), 7.75 (d, 1H, J_{5,6}=8 Hz, H-6), 7.40-6.70 (m, 13H, ArH), 5.59 (d, 1H, J_{5,6}=8 Hz, 5-H), 5.28 (s, 2H, 0-CH₂-N), 4.65 (br s, 1H, OH), 4.45 (d, 1H, CH); 3.71 (s, 6H, OCH₃), 3.40 (m, 2H, CH₂-CH-CH₂), 3.00 (m, 2H, CH₂-CH-CH₂).

Compound (<u>1b</u>) (640 mg, 2.7 mmol) was treated with dimethoxytrity1 chloride (1.01 g, 2.97 mmol) in dry pyridine (54 ml) at room temperature for 5 h, and the mixture was processed in the same manner as described above giving <u>2b</u> (816 mg, 57 %); mp 82-86°C; R_f 0.23 (solvent A); UV (Me-OH) λ_{max} 267, 237 nm, λ_{min} 253 nm; ¹H-NMR (DMSO-d₆) & 10.83 (br s, 1H, NH), 8.20 (d, 1H, J_{5,6}=7 Hz, H-6) 7.37-6.74 (m, 13H, ArH), 5.20 (d, 2H, O-CH₂-N), 4.45 (br s, 1H, OH), 3.75 (s, 6H, OCH₃), 3.45 (m, 3H, CH₂-CH-CH₂), 3.00 (m, 2H, CH₂-CH-CH₂).

Silylation of la. Compound (la) (362 mg, 1.68 mmol) and imidazole (137 mg, 2.1 mmol) were dissolved in DMF (12 ml). TBDMS-Cl 10 (254 mg, 1.85 mmol) was added in portions over 20 min period and the mixture was stirred for an additional 30 min. The solvent was removed at reduced pressure and the residue was chromatographed on a silica gel column using CH $_2$ Cl $_2$ -MeOH (95:5, v/v) as eluant. The appropriate fractions were evaporated in vacuo to give the silylated product (2c) (260 mg, 47 %); mp 86-88°C; R $_f$ 0.20 (solvent A); UV (MeOH) $\lambda_{\rm max}$ 256 nm, $\lambda_{\rm min}$ 225 nm; 1 H-NMR (CDCl $_3$) δ 11.20 (br s, 1H, NH), 7.60 (d, 1H, J $_5$, 6=7.5 Hz, H-6), 5.55 (d, 1H, J $_5$, 6=7.5 Hz, H-5), 5.15 (s, 2H, O-CH-N), 4.60 (br s, 1H, OH), 3.45 (m, 5H, CH $_2$ -CH-CH $_2$), 0.85 (s, 9H, (CH $_3$) $_3$ C), 0.00 (s, 6H, (CH $_3$) $_2$ Si); Anal Calcd for C $_1$ 4 $_2$ 6 $_2$ 0 $_2$ 5si : C, 51.07; H, 8.18; N, 8.33 %, Found : C, 50.87; H, 7.93; N, 8.47 %.

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22°C for 30 min. After the precipitated triethylammonium hydrochloride was removed by filtration, the filtrate was evaporated in vacuo to give a colorless liquid. The crude material was fractionally distilled from cesium fluoride to afford 5.6 g (65 %) of 3; b.p. 77-80°C/3 mmHg; 1 H-NMR (CDCl₃) $_{\delta}$ 4.5 (m, 2H, CH₂-0), 3.5-2.9 (m, 2H, CH₂-N), 2.60 (d, 3H, 1 J_{H-P}=16 Hz, NCH₃); 31 P-NMR (CDCl₃, 85 % H₃PO₄) $_{\delta}$ 167.

Phosphorylation of 2. A solution of 2a (1.22 g, 2.35 mmol) in dry THF (9.4 ml) and disopropylethylamine (0.60 ml, 3.52 mmol) was cooled at -50°C and compound (3) (486 mg, 3.52 mmol) was added with vigorous stirring. After 30 min, the precipitated diisopropylethylammonium hydrochloride was removed by filtration and the filtrate was treated with MCPBA (603 mg, 3.52 mmol) at 20°C. After 15 min, pyridine was added, and the solvent was evaporated in vacuo. The residue was purified by chromatography on a column (3 x 10 cm) of silica gel which was eluted with ethyl acetate-ethanol-pyridine (8:1:1, v/v). The appropriate fractions were collected and the solvent was evaporated under reduced pressure to give the product (5a) which was isolated as a white powder (1.14 g, 79 %) by precipitation from hexane (100 m1); mp 69-72°C; UV (MeOH) $\lambda_{\rm max}$ 265 (sh), 233 nm; 1 H-NMR (pyridine-d₅) δ 7.45-7.05 (m, 14H, H-6 and ArH), 5.80 (d, 1H, $J_{5,6}$ =7 Hz, H-5), 5.50 (br s, 2H, O-CH₂-N), 4.20 (m, 2H, CH₂-O), 3.70 (s, 6H, OCH₃), 3.60-2.95 (m, 7H, CH_2-N and $CH_2-CH-CH_2$), 2.55 (d, 3H, $J_{H.P} = 10 \text{ Hz}$, NCH_3); $^{31}P-NMR$ (pyridine-d₆, 85 % H_3PO_4) δ 22.63, 22.29. Anal Calcd for C₃₂H₃₄N₃O₀P.3.5 H₂O : C, 55.00; H, 5.92; N, 6.02 %. Found : C, 54.99; H, 5.83; N, 6.00 %.

Compound 5b was prepared by a similar procedure as compound 5a, except that the acyclic nucleoside derivative 2b (1.05 g, 2.0 mmol) was treated with 3 (413 mg, 3.0 mmol). 5b was isolated as a white powder (950 mg, 76 %); mp 73-76°C; UV(MeOH) $\lambda_{\rm max}$ 261, 227 nm, $\lambda_{\rm min}$ 247 nm; 1 H-NMR (pyridine- d_5) δ 7.60-6.70 (m, 14H, H-6 and ArH), 5.27 (br s, 2H, 0-CH₂-N), 4.25 (m, 2H, CH₂-0), 3.73 (s, 6H, 0CH₃), 3.60-2.85 (m, 7H, CH₂-N and CH₂-CH-CH₂), 2.72 (d, 3H, J_{H,P}=10 Hz, N-CH₃); 31 P-NMR (pyridine- d_5 , 85 % H₃PO₄) δ 21.95, 21.81. Anal Calcd for C_{32} H₃₃O₉N₃PF.H₂O : C, 57.20; H, 5.25; N, 6.25 %. Found : C, 57.14; H, 5.17; N, 6.13 %.

Compound $\underline{5c}$ was prepared by a similar procedure as compound $\underline{5a}$ except that the acyclic nucleoside derivative $\underline{2c}$ (661 mg, 2.0 mmol) was treated with $\underline{3}$ (413 mg, 3.0 mmol). $\underline{5c}$ was isolated as a white powder (707 mg, 79 %); UV (MeOH) λ_{\max} 255 nm; $\frac{1}{1}$ H-NMR (CDCl $_3$) δ 9.74 (br s, 1H, NH),

7.47 (d, 1H, $J_{5,6}$ =8 Hz, H-6), 5.74 (d, 1H, $J_{5,6}$ =8 Hz, H-5), 5.26 (s, 2H, 0-CH₂-N), 4.20 (m, 2H, CH₂-O), 4.05-3.60 (m, 5H, CH₂-CH-CH₂), 3.30-3.05 (m, 2H, CH₂-N), 2.60 (d, 3H, $J_{H,P}$ =10 Hz, NCH₃), 0.80 (s, 9H, (CH₃)₃C), 0.00 (s, 6H, (CH₃)₂Si); 31 P-NMR (CDCl₃, 85 % H₃PO₄) & 21.74, 21.63. Anal Calcd for C_{17} H₃₀N₃O₇PSi : C, 45.62; H, 6.76; N, 8.80 %. Found : C, 45.73; H, 6.89; N, 8.65 %.

Removal of the trityl groups from 5a,b. Compound 5a (420 mg, 0.68 mmol) was treated with $\rm ZnBr_2$ (1.53 g, 0.68 mmol) in dry $\rm CH_3NO_2$ (17 ml) at 0°C for 15 min. The solvent was concentrated and applied onto a silica gel column (2 x 8 cm). The appropriate fractions [eluted with ethyl acetate-ethanol-pyridine (70:25:5, v/v)] were collected and the solvent was evaporated under reduced pressure to give the product (6a) as a white powder (192 mg, 85 %); mp 68-71°C; UV (MeOH) $\lambda_{\rm max}$ 261 nm; 1 H-NMR (pyridine-d₅) δ 10.52 (br s, 1H, NH), 7.35 (d, 1H, J_{5,6}=7 Hz, H-6), 5.64 (d, 1H, J_{5,6}=7 Hz, H-5), 5.23 (br s, 2H, 0-CH₂-N), 4.50 (br s, 1H, OH), 4.06-3.36 (m, 9H, CH₂-CH-CH₂ and 0-CH₂-CH₂-N), 2.87 (d, 3H, J_{H,P}=10 Hz, NCH₃).

Compound <u>6b</u> was prepared by a similar procedure as compound <u>6a</u>, except that the phosphorylated product (<u>5b</u>) (700 mg, 1.12 mmo1) was treated with ZnBr_2 . <u>6b</u> was isolated as a white powder (305 mg, 81 %); mp 66-68°C; UV (MeOH) λ_{max} 265 nm; ¹H-NMR (pyridine-d₅) δ 11.02 (br s, 1H, NH), 7.27 (d, 1H, $J_{5,6}$ =8 Hz, H-6), 5.33 (m, 2H, O-CH₂-N), 4.51 (br s, 1H, OH), 4.17-3.40 (m, 9H, CH₂-CH-CH₂ and O-CH₂-CH₂-N), 2.85 (d, 3H, $J_{\text{H,P}}$ =10 Hz, NCH₃).

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